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Physics Chapter Properties Of Matter

Everything around us has mass and volume and they occupy space, and we called them as matter. It can be in four state, like solid, liquid, gas and plasma. We will talk about main properties of matter in this unit like, mass, volume, density, elasticity, inertia...Etc. You can classify matters with their physical or observable properties and chemical or unobservable properties, for example their smells, colors, shapes give you an idea about it.

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Thermal Properties of Matter Class 11 Notes Physics Chapter 11 • Heat is the form of energy transferred between two (or more) systems or a system and its surroundings by virtue of... • Temperature of a substance is a physical quantity which measures the degree of hotness or coldness of the ...

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Read Online Physics Chapter Properties Of Matter Answers Physics Chapter Properties Of Matter Everything around us has mass and volume and they occupy space, and we called them as matter. It can be in four state, like solid, liquid, gas and plasma. We will talk about main properties of matter in this unit like, mass,

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Chapter 10 - Thermal Properties of Matter

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One kilogram of ice at 0°C is mixed with one kilogram of water at 80°C . The final temperature of the mixture is (Take: Specific heat of water = $4200\text{ J kg}^{-1}\text{K}^{-1}$, Latent heat of ice = 336 kJ kg^{-1})

(1) 0°C (2) 50°C (3) 40°C (4) 60°C

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THERMAL PROPERTIES OF MATTER
The branch dealing with
measurement of temperature is
called thermometry and the devices
used to measure temperature are
called thermometers.

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THERMAL PROPERTIES OF MATTER
Properties Words that describe
matter (adjectives) Physical
Properties- a property that can be
observed and measured without
changing the material ' s
composition. Examples- color,
hardness, m.p., b.p. 10.

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The thermal expansion of solids is the least among the three states of matter. Explain. (a) Arrange solids, liquids and gases in increasing order of the following properties of matter

 rigidity (ii) diffusion (iii) compressibility.
 (b) Write one example from your daily life which is based on diffusion of gases.

Properties Of Matter - Solids - Problems

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